19-Norclerodane Diterpenoids

The Total Synthesis of Racemic Teucvin and 12-epi-Teucvin**

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During the past three decades, over one thousand clerodane (1) diterpenoids, including some 19-nor variants, have been isolated. Owing to their interesting structural features and the wide range of potentially useful biological activities detected for many of the relatively few compounds tested, extensive effort has been directed towards the synthesis of clerodanoids in recent years. Use report herein the first total synthesis, in racemic form, of teucvin (2), an amoebi-

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[**] We are grateful to the National Science Council of the Republic of China and the Natural Sciences and Engineering Research Council of Canada for financial support.

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cidal agent and root growth inhibitor^[4] that was first isolated in 1973 from *Teucrium viscidum* Blume by Fuijta and coworkers,^[3a] and 12-*epi*-teucvin (3),^[5] a congener of teucvin isolated eleven years later also from a *Teucrium* species by Rodriguez and Savona, et al. These 19-norclerodane diterpenoids are of considerable structural complexity. In addition to the presence of five stereogenic centers, challenging structural features in 2 and 3 comprise a decalin core attached to an α , β -unsaturated- γ -lactone unit and a spiro γ -lactone moiety containing a pendant furyl group.

The present work began with the synthesis of dienophile **4** (Scheme 1). Its Diels–Alder addition to *trans*-2,4-pentadien-

Scheme 1. Synthesis of dienophile **4.** a) NaH, EtOCHO, EtOH (cat.), THF, RT, 10 h, 90%; b) CH₂=CHCOCH₃, DABCO, THF, RT, 10 h, 86%; c) p-TsOH, PhH, reflux, $-H_2O$, 8 h, 89%; d) NaH, EtOCHO, EtOH (cat.), THF, RT, 10 h, 89%; e) DDQ, K_2CO_3 , THF, 0°C, 20 min, 90%.

1-ol (see Scheme 2), which provided a rapid access to the required decalin core with suitable functionalities at strategic positions, served as a key operation to tackle the synthetic problems associated with the title compounds. Dienophile 4 was prepared from diethyl succinate in five steps in an overall vield of 55% (Scheme 1). Formylation of diethyl succinate with ethyl formate in THF in the presence of sodium hydride and a small amount of ethanol gave the corresponding aldehyde 5, which was subjected to Michael addition with methyl vinyl ketone using 1,4-diazabicyclo[2.2.2]octane (DABCO) as a base to give the keto aldehyde 6. The aldol condensation of 6 was subsequently induced under acidic conditions using p-toluenesulfonic acid (p-TsOH). The resulting enone 7 was formylated with ethyl formate as before to give the formyl enone 8. Finally, the dienophilic double bond was introduced by dehydrogenation with 2,3dichloro-5,6-dicyano-1,4-benzoquinone (DDQ) to give the desired dienophile 4.

Under zinc chloride catalysis, dienophile **4** was found to undergo Diels–Alder reaction rapidly (0 °C, 1 h, in CH_2Cl_2) with *trans*-2,4-pentadien-1-ol^[6] to give compound **9** in a 83 % yield, as a result of concomitant intramolecular hemiacetal formation, together with a small amount (≈ 7 %) of the deformylation by-product **10** (Scheme 2). Before the outset of the current work, we had observed that the placement of an ester group at C4 of various cyclohexenone dienophiles could induce a profound effect on the face-selectivity of the Diels–

Scheme 2. Synthesis of target compounds **2** and **3**. a) ZnCl₂, *trans*-2,4-pentadien-1-ol, CH₂Cl₂, 0°C, 1 h, 90% (**9**:10=11:1); b) Ac₂O, py, DMAP (cat.), RT, 10 h, 90%; c) (CH₃)₂CuLi, Et₂O, 0°C, 0.5 h, 82%; d) NaOH, EtOH, H₂O, RT, 10 h, 90%; e) H₂ (60 psi), 10% Pd/C, EtOAc, RT, 6 h, 89%; f) Jones reagent, acetone, 0°C, 1 h, 90%; g) *p*-TsOH, PhH, reflux, 5 h, 85%; h) (COCl)₂, PhH, reflux, 10 min; i) LiAl (tBuO)₃H, THF, -40°C, 1 h, 70% over two steps; j) 3-lithiofuran, Et₂O, -78°C, 20 min; k) LiH, THF, RT, 10 h, 45% over two steps (**2**:**3**=1:3).

Alder reaction; the addition of the diene was found to take place specifically from the side containing the ester group. To our delight, this interesting phenomenon of face-control, central to our synthetic design, was also realized experimentally in the Diels—Alder addition of *trans*-2,4-pentadien-1-ol to compound 4. This reaction occurred in a completely face-selective manner in favor of the face containing the carbethoxy group, thus allowing full control of the two crucial stereogenic centers (C9 and C10 (clerodane numbering)). This key reaction also proceeded with complete stereo- (*cis*-addition and *endo*-addition with respect to the ketone carbonyl group) and regioselectivity (*ortho*-addition); both the observed products originated from the same adduct whose structure was unequivocally established after further transformations (vide infra).

For the incorporation of the required methyl group at C17, the hemiacetal group of **9** was protected as the acetate by using acetic anhydride, pyridine, and a small amount of 4-dimethylaminopyridine (DMAP). The resulting acetate **11** (90% yield) was subsequently treated with lithium dimethyl cuprate. The conjugate addition took place preferentially

from the sterically less hindered side to give ketone **12** as the predominant product (82% yield) along with a minor quantity (13% yield) of the epimeric addition product. The structure of **12** was unambiguously confirmed by an X-ray crystallographic analysis.^[8]

The protected acetal group present in compound 12 was removed by treatment with aqueous sodium hydroxide in ethanol. Under these conditions, the more exposed ester group was also selectively hydrolyzed to give acid 13 in 90% yield. A single isomer was obtained and its trans-ring junction was induced as a result of thermodynamic control. Catalytic hydrogenation of 13 using 10 % Pd/C as a catalyst gave rise to the corresponding saturated compound 14, which was then oxidized with Jones reagent to give diacid 15 in 80% yield over two steps. The bridged γ-lactone ring was introduced by enol lactone ring formation followed by isomerization of the double bond. This was effected by treatment with p-toluenesulfonic acid in refluxing benzene. Under these thermodynamically controlled conditions, the more stable epimer 16 was obtained as the exclusive product in 85 % yield. To install the spiro lactone ring, acid 16 was treated with oxalyl chloride to produce the corresponding acid chloride which, without purification, was immediately reduced with lithium aluminum tri-tert-butoxy hydride to give aldehyde 17 in 70% yield. This was followed by addition of 3-lithofuran to give a mixture of two epimeric alcohols in a ratio of 3:1 as determined by the ¹H NMR analysis. These alcohols, which were found to be rather labile and deteriorated readily upon column chromatography, were immediately subjected to lactonization with lithium hydride in THF, and racemic teucvin (2; m.p. 172-174°C) and 12-epi-teucvin (3; m.p. 201-203°C) were produced in a 1:3 ratio and in a total yield of 45% from 17. The ¹H NMR, ¹³C NMR, and mass spectral data obtained for the synthetic materials are in good agreement with those reported for the naturally occurring compounds.[3,5] Thus, the synthetic challenges presented by the title compounds have been successfully met by employing an intermolecular Diels-Alder approach that is rather efficient in terms of stereo- and regiochemical control, the number of steps involved (sixteen steps), and the overall yield (1.6% for teucvin and 4.8% for 12-epi-teucvin).

Received: December 2, 2002 [Z50681]

Keywords: Diels-Alder reaction \cdot natural products \cdot terpenoids \cdot teucvin \cdot total synthesis

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